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Dedicated, in admiration, to Professor Teruaki Mukaiyama on the occasion of his 80th birthday

The title compounds, **4** and **7**, have been prepared from the corresponding α -amino acid derivative selenocystine (**1**) by the following sequence of steps: cleavage of the Se–Se bond with NaBH₄, *p*-methoxybenzyl (PMB) protection of the SeH group, Fmoc or Boc protection at the N-atom and *Arndt– Eistert* homologation (*Schemes 1* and 2). A β^3 -heptapeptide **8** with an N-terminal β^3 -hSec(PMB) residue was synthesized on *Rink amide AM* resin and deprotected ('in air') to give the corresponding diselenide **9**, which, in turn, was coupled with a β^3 -tetrapeptide thiol ester **10** by a seleno-ligation. The product β^3 -undecapeptide was identified as its diselenide and its mixed selenosulfide with thiophenol (*Scheme 3*). The differences between α - and β -Sec derivatives are discussed.

1. Introduction. – Selenium is an essential trace element in human nutrition [1]. Recent studies have also reported beneficial effects of selenium compounds for cancer prevention [2]. Among selenium derivatives, selenocysteine (Sec or U), the 21st proteinogenic amino acid [3], has been the focus of considerable attention³). This moiety is found in a variety of naturally occuring enzymes (selenoproteins) such as glutathione peroxidase, iodothyronine deiodinase, and thioredoxin reductase [4]. The basis for the incorporation of Sec into proteins has been established by *Böck*'s pioneering work on *Escherichia coli* mutants [5]. Selenocysteine is incorporated into selenoproteins by a process that decodes the UGA codon as selenocysteine. For most human selenoproteins, 'homologues' exist, in which selenocysteine is replaced by cysteine. These enzymes are normally weaker catalysts when compared to the corresponding selenoproteins [6]. The selenol group ($pK_a \approx 5.7$) is more acidic than the thiol group ($pK_a \approx 8.5$) [7]. Thus, in contrast to cysteine, selenocysteine is ionized at physiological pH. Furthermore, the Se-atom has a higher polarizability, which makes the selenol group a 'softer' S_N^2 -type nucleophile.

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²) Part of the projected Ph.D. thesis of G.C., ETH-Zürich.

³) For a comprehensive review 'From Selenium to Selenoproteins: Synthesis, Identity, and Their Role in Human Health' and for a discussion on 'The Importance of Selenium to Human Health' see [2b,c].

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Diselenides have lower reduction potential as compared to disulfides, and selenocysteine-containing peptides form diselenides more rapidly than the corresponding cysteine-containing peptides form disulfides [8]. The presence of selenium in an amino acid sequence provides a spectroscopic as well as a mechanistic probe of peptide properties⁴).

Furthermore, the selenol group of selenocysteine can be easily eliminated under mild oxidative conditions to form dehydroalanines, which are convenient precursors for the preparation of peptide conjugates [10]. For these reasons, interest in selenocysteine and its derivatives has dramatically increased in recent years.

Many different syntheses of selenocysteine derivatives have been reported [11]. In a common approach, a selenolate acts as nucleophile on an activated OH group of a serine analog. For solid-phase peptide synthesis (SPPS) *N*-Boc- or *N*-Fmoc-protected derivatives are the most convenient. The currently most useful protective group for a selenol is the *para*-methoxybenzyl-group (PMB or Mob group) [11b,c]. The PMB group can be removed under various conditions. For instance, *Fujii* and co-workers used oxidative conditions (I₂ in AcOH or DMSO in CF₃COOH (TFA)) or strong *Lewis* acids (Me₃SiBr (TMSBr) or TMSOTf (Tf = trifluoromethylsulfonyl)) [11b]. An alternative method utilizes 2-nitrophenylsulfenyl chloride [12] in analogy to protocols for deprotection of related cysteine derivatives [13][14]. A milder method for deprotection of Sec(PMB) has been recently reported by *Hondal* and co-workers [15]: best results were obtained using 2,2'-dithiobis(5-nitropyridine) (DTNP).

Consequently, typical building blocks for SPPS of Sec-containing peptides are Fmoc-Sec(PMB)-OH and Boc-Sec(PMB)-OH. However, special care must be taken while performing SPPS with these derivatives, since they are prone to racemization and elimination under basic conditions. In particular, iterative piperidine deprotection steps can lead to the formation of dehydroalanine moieties within the peptide and subsequently to the corresponding piperidine adducts [16][17] (replacement of CH_2SeR by $CH_2-N(CH_2)_5$). To minimize these problems the incorporation of selenium residues is usually performed with preformed pentafluorophenyl esters (OPfp esters) in the presence of 1-hydroxy-1*H*-benzotriazole (HOBt) [16][17], which obviates the need for tertiary amines.

For the synthesis of larger peptides, *Kent*'s native chemical ligation, the coupling of two unprotected peptides, has proven to be a useful tool [18]. In this convergent synthetic strategy, a C-terminal thioester group of one peptide is coupled with an N-terminal cysteine residue of another peptide. This technique has also been extended to the coupling of a C-terminal thioester with an N-terminal Sec moiety⁵). In contrast to the conventional native chemical (thio)ligation, where disulfide bonds are cleaved with thiophenol, a phosphine (usually tris(2-carboxyethyl)phosphine (TCEP)) is often added for reductive cleavage of diselenide bonds in the reaction mixture of a (seleno)ligation [19][20]. In some cases, though, thiophenol is sufficient for this purpose [11c][21][22]. Expressed protein ligation [23] has been used to prepare

⁴) A general overview on the replacement of sulfur by other chalcogens in peptides and proteins is given in [9].

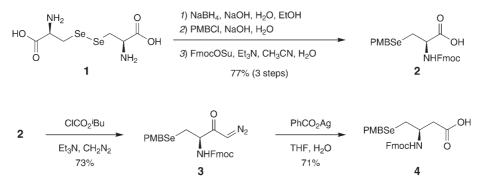
⁵) An overview of the use of Sec derivatives for chemoselective ligations is given in [19].

selenoproteins [20] [24]; here, the thioesters are generated by using recombinant DNA technology and condensed with a synthetic Sec-containing segment.

β-Peptides represent a biomedically promising class of peptidomimetics. Like their *α*-counterparts, β-peptides fold to secondary structures such as turns, helices, or sheets [25]. Furthermore, they can mimic *α*-peptidic hormones, as well as peptide – protein and protein – protein interactions [25a]. They may be orally bioavailable [26], and, most importantly, they are enzymatically and metabolically stable [25a]. The β^2 - and β^3 -amino acid homologues of the 20 proteinogenic *α*-amino acids have all been prepared and incorporated in β-peptides, the properties of which have been studied extensively [25a]. Since the 21st proteinogenic amino acid, selenocysteine (Sec), confers interesting features to *α*-peptide synthesis. We here describe the preparation of Fmoc- β^3 Sec(PMB)-OH and of Boc- β^3 Sec(PMB)-OH, as well as their use in solid phase synthesis and in native chemical ligation.

2. Preparation of Fmoc- β^3 hSec(PMB)-OH and Boc- β^3 hSec(PMB)-OH. – Fmoc- β^3 hSec(PMB)-OH (4) was synthesized starting from commercial selenocystine (1). The Se–Se bond of 1 was cleaved with NaBH₄, and the selenol was protected with a PMB group, according to the procedure published by *Fujii* and co-workers [11b], and *Vermeulen* and co-workers [11e]. Subsequent treatment with *N*-({[(9*H*-fluoren-9-yl)methoxy]carbonyl}oxy)succinimide (Fmoc-OSu) furnished Fmoc-Sec(PMB)-OH (2) [11b] (*Scheme 1*) in 77% yield over three steps.

Scheme 1. Preparation of $Fmoc-\beta^3 hSec(PMB)-OH(4)$. For abbreviations, see accompanying text.



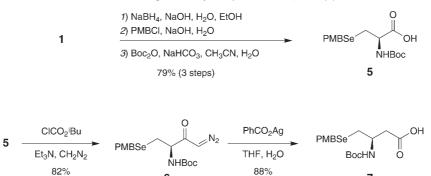
Homologation to the β -amino-acid derivative **4** was accomplished by the *Arndt*– *Eistert* protocol [25d][27][28], wherein the diazo ketone **3** was converted to **4** by an ultrasound-promoted [29] *Wolff* rearrangement [30] with PhCOOAg.

Boc- β^3 hSec(PMB)-OH (7) was obtained in the same way as the Fmoc derivative (*Scheme 2*). Again, the corresponding α -amino acid (α -Boc-Sec(PMB)-OH, (5)) was first prepared, and homologation through the diazo ketone 6 led to the desired *N*- and *Se*-protected carboxylic acid 7.

According to these simple protocols, $\text{Fmoc}-\beta^3\text{hSec}(\text{PMB})$ -OH **4** and Boc- $\beta^3\text{hSec}(\text{PMB})$ -OH **7** can be prepared in five steps starting from commercial precursors in 40 and 57% overall yield, respectively.

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Scheme 2. Preparation of $Boc-\beta^3hSec(PMB)-OH(7)$

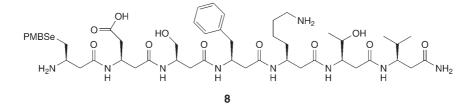


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Due to the high nucleophilicity of selenium compounds of the general formula R-Se-R, we had expected difficulties in the reaction with CH_2N_2 ($2 \rightarrow 3$ and $5 \rightarrow 6$) and in the *Wolff*-rearrangement step of the intermediate diazo ketone ($3 \rightarrow 4$ and $6 \rightarrow 7)^6$)⁷). However, the overall yield of diazo-ketone formation and *Wolff* rearrangement ($2 \rightarrow 4, 5 \rightarrow 7$) was comparable to that of the corresponding sulfur derivative (Fmoc-Cys(PMB)-OH or Fmoc-Cys(Acm)-OH \rightarrow Fmoc- β^3h Cys(Trt)-OH) [28c]).

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3. Preparation of a β -Homoselenocysteine-Containing β -Peptide. – Boc- β^3 hSec(PMB)-OH was used for the preparation of the β -peptide 8, which, in turn, was used for a (seleno)ligation to the undecapeptide 11 (*cf. Scheme 3*, below).



As mentioned in the *Introduction*, in α -peptide synthesis, incorporation of Sec is usually performed with the corresponding pentafluorophenyl esters and HOBt

- 6) After all, selenoethers, more so than thioethers, are expected to react with a carbene to form selenoylides [31]: *cf.* the catalytic enantioselective oxirane formation by *in-situ* trapping of a carbene (generated from a diazo compound) with a thioether and subsequent reaction with an aldehyde [32].
- ⁷) This may suggest that no free carbene is involved in the *Wolff* rearrangement of **3** and **6**, compatible with the view that migration and loss of nitrogen are concerted [33][34]:

$$Pg_{Se} \xrightarrow{(O)}_{NH \to H} N_{2}^{\oplus} \xrightarrow{(-N_{2})} N_{2}^{\oplus} \xrightarrow{(-N_{2})} N_{2}^{H} \xrightarrow{(-N_{2})} N_{2}^{H} \xrightarrow{(-N_{2})} 4,7$$

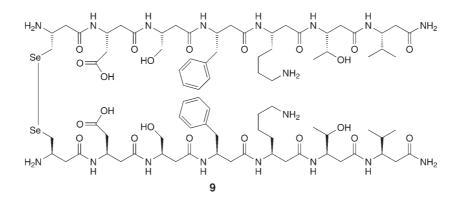
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catalysis [16] to avoid racemization and elimination [17]. Neither of these processes is expected to occur with β -homoselenocysteine, since the side chain is in the β - and not in the α -position of the carbonyl group. For the synthesis of the β -heptapeptide **8** with an N-terminal β ³hSec residue, we used conventional solid-phase coupling on a *Rink-Amide AM* resin. The couplings were achieved with HATU and Hünig's base in DMF. Prior to the incorporation of the β -homoselenocysteine moiety, the resin-bound β peptide was divided into two equal portions.

The Boc (PMB-protected) β -homoselenocysteine was coupled to the first half of the resin-bound peptide by standard methods (HATU and DIPEA as tertiary base). The use of *N*-Boc-protected β -selenocysteine obviates the need for a final, N-terminal Fmoc deprotection. After this last coupling step, the peptide was removed from the resin under conditions that did not set the selenol groups free. After purification by preparative HPLC, the heptapeptide **8** was isolated as its TFA salt in a disappointingly low 21% yield.

Alternatively, β -homoselenocysteine was attached to the other half of the resinbound peptide by activation as its pentafluorophenyl (PFP) ester⁸) in the presence of HOBt. After cleavage from the resin and purification by preparative HPLC, the heptapeptide **8** was isolated as its TFA salt in a similarly low 19% yield.

When samples of the heptapeptide **8** were treated with a mixture of *m*-cresol, thioanisol, TFA, and TMSBr (50:120:750:132) for 1 h at 0°, followed by preparative HPLC purification, the diselenide **9** was isolated as its TFA salt in 59% yield⁹).



Thus, both methods of attaching the β^3 hSec to the resin-bound β -hexapeptide afforded the β -heptapeptide **8** (in comparable overall yield). As expected, base treatment does not seem to have an adverse effect on the results in the case of β^3 hSec derivatives.

⁸) Boc- β^3 hSec(PMB)-OPFP was prepared according to [35].

⁹⁾ Additionally, 18% starting material 8 was recovered.

4. Selenoligation. – Coupling of the β -heptapeptide fragment derivative **9** with the β -tetrapeptide thiol ester¹⁰) **10** was achieved by Sec-mediated chemical ligation (selenoligation [11c][20][21]) as depicted in *Scheme 3*.

The TFA salts of the lyophilized peptides were transferred to a *Schlenk* flask, and degassed phosphate buffer (100 mM, pH 8.5) containing 6M GdmCl was added to dissolve them. After adjusting the pH of the solution to 7.5, the reaction was initiated with 5% thiophenol (v/v) under N₂. The course of the reaction was monitored by HPLC and LC/MS analyses. The conversion was judged to be complete after 8 h¹¹). The disappearance of the starting materials was accompanied by formation of two new compounds (*Fig. 1*): the faster moving species corresponded to the diselenide **11c** of peptide **11a** ([M + H]⁺, C₁₃₆H₂₀₅N₂₆O₃₂Se⁺₂; found: 2874.0; calc. 2874.3 Da)¹²), whereas the slower moving species had a mass which is compatible with the selenosulfide **11b** formed from peptide **11a** and thiophenol ([M + H]⁺, C₇₄H₁₀₈N₁₃O₁₆SSe⁺; found: 1546.8; calc. 1546.7 Da).

After completion of the reaction, the ligation mixture was treated with dithiothreitol (DTT; 4 mg/ml reaction) for 30 min under N₂ in order to generate a homogenous solution that was subsequently washed twice with Et₂O. The resulting aqueous solution was diluted with 3 ml of MeCN/H₂O 1:1 containing 0.1% of TFA, and then directly injected into the preparative HPLC column. The diselenide product **11c** was isolated as the TFA salt in 50% yield and characterized by analytical HPLC and high-resolution MALDI-MS (*Fig. 2*).

The normalized CD spectra in MeOH of the β -peptidic diselenides 9 and 11c, and of the mixed selenosulfide 11b, are shown in *Fig. 3*. The starting diselenide 9 and the selenosulfide product 11b exhibit the characteristic *Cotton* effect of a 3_{14} -helix: a minimum around 214 nm and a maximum near 200 nm. The diselenide 11c of the ligation product shows, instead, a minimum near 208 nm but no positive *Cotton* effect at shorter wavelength.

5. Discussion and Conclusions. – The Arndt–Eistert-homologation sequence of steps is readily applicable to selenocysteine (H-Sec-OH) for the preparation of Boc or Fmoc protected β^3 hSec derivatives in good yields (57 and 40%, resp., for five steps from selenocystine). Furthermore Fmoc- β^3 hSec(PMB)OH was successfully employed for selenoligation of a β -hepta- and a β -tetrapeptide to give derivatives **11b** and **11c** of the β -undecapeptide **11a**.

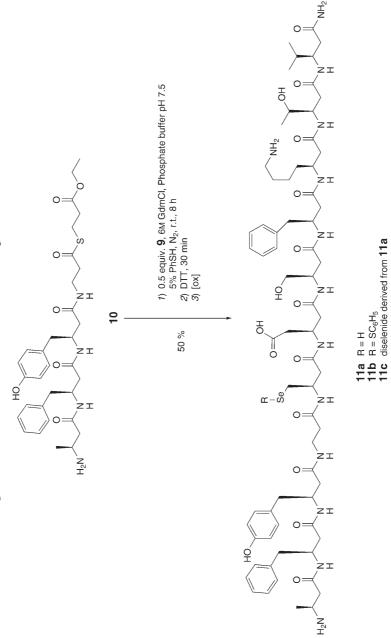
As noted in the *Introduction*, Sec residues are convenient precursors of the dehydroalanine moiety **A**, the C=C bond of which is captodatively [37] substituted. Analogous elimination with the β^3 hSec residue would lead to a dehydro- β^3 hAla unit **B** in which the C=C bond is substituted by an acylamido and a CH₂CO group. The C=C bond in **B** is expected to be a better donor for electrophiles than that in **A**. The product of electrophilic attack on **B**, an acyliminium ion **C**, is likely to lose a proton from the α -

¹⁰) The preparation of the thioester **10** is described in [36]. We thank Dr. *Thierry Kimmerlin* for the preparation of the sample we used (*cf.* ETH Dissertation No. 15800, p. 83, Zürich, 2004).

¹¹) For a comparison HPLC chromatograms after 0, 5 and 18 h see Fig. 1.

¹²) The masses reported in the discussion were obtained from LC/ESI-MS. The exact masses are reported in the *Exper. Part.*







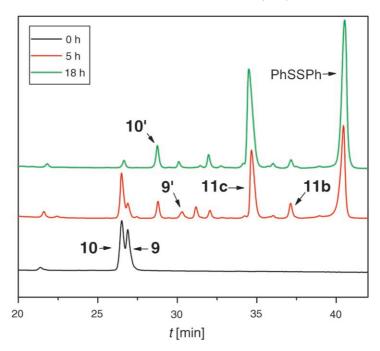
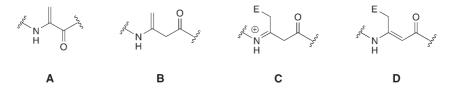


Fig. 1. Analytical HPLC traces of the ligation reaction between diselenide 9 and thioester 10. At time zero (without thiophenol), the two starting materials are visible. Five hours after PhSH addition, the coupling products 11b and 11c accumulated. After 18 h, the starting diselenide 9 was almost completely consumed, and the thiophenol ester 10' (SPh instead of $S(CH_2)_2CO_2Et$ in 10) as well as the 'mixed' selenosulfide 9' (PhSSe derivative formed from 9) were identified by LC/ESI-MS. Chromatographic conditions: C_8 column (Macherey-Nagel 250 mm × 4.6 mm × 300 Å, 5 µ; 5–95% B in 60 min: 5 min 5% B, 50 min 60% B, 55 min 95% B).

carbonyl position to give an α,β -unsaturated β -acylamino-carbonyl system **D** ('vinylogous amide'). With E = H in **C** and **D**, this process would be a simple double-bond shift, and, with other electrophiles E, there would be a possibility of elaborating the structure further.



The CD spectrum of the diselenide **11c** (green curve in *Fig. 3*), as compared to those of the β -peptides **9** and **11b**, provides yet another example of the structural changes that are seen when long-chain β -peptides are investigated. Based on an analogy to many structurally characterized β -peptides that exhibit a negative *Cotton* effect near 215 nm and a positive *Cotton* effect near 200 nm, the CD spectra shown in *Fig. 3* suggest that β -peptides **9** and **11b**, which contain 14 and 11 β -amino-acid residues,

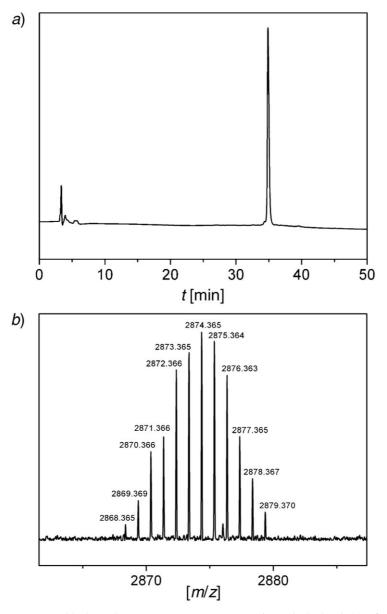


Fig. 2. HPLC Trace and high-resolution MALDI mass spectrum of peptide diselenide **11c**. a) Analytical HPLC chromatogram of purified **11c**; conditions: C_8 column (Macherey-Nagel 250 mm × 4.6 mm × 300 Å, 5 µ; 5–95% B in 60 min: 5 min 5% B, 50 min 60% B, 55 min 95% B). b) High resolution MALDI of **11c**. Shown is the isotopic pattern of $[M + H]^+$ between the sixth and the seventeenth isotopic peak. A value of 2874.365 was calculated for the twelfth isotopic peak based on a full width at half maximum resolution of 30,000 using the Exact Mass Calculator program from the IonSpec Corporation. The mass found for the corresponding peak was 2874.359.

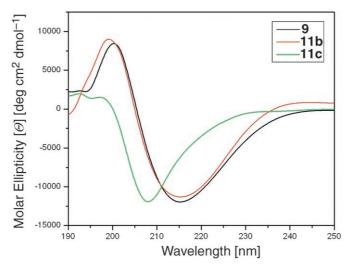


Fig. 3. Normalized CD Spectra of 9 and 11. The spectra were recorded in MeOH, at 20°, with 0.25 mm 9, 0.19 mm 11c, and 0.18 mm 11b. Solutions were prepared by weighing the peptide on a µg-precision balance and dissolving it in a volumetric flask.

respectively, adopt 3_{14} -helical structures [25]. Hypsochromic shifts of the negative and the 'loss' of the positive *Cotton* effect, as seen in the CD spectrum of the diselenide **11c**, have been observed previously with larger β -peptides [25a,b,e][36][38] but never interpreted. As we and others have noted [25e][38], assignment of β -peptidic secondary structure based on CD data alone must be done with considerable caution¹³). In the present case, the spectral changes observed upon dimerization of the β -peptide might be due to aggregation, but in the absence of detailed NMR studies other explanations cannot be ruled out.¹⁴) Aggregated peptides often show CD spectra, that are different from those of the corresponding monomers. A good example is the equilibrium between non-helical monomers that form ordered four-helix bundles [39].

Experimental Part

1. General. Abbreviations: Bn: benzyl, Boc: (tert-butoxy)carbonyl, DIPEA: diisopropylethylamine $(EtN(i-Pr)_2)$, DMAP: 4-(dimethylamino)pyridine, DMF: dimethylformamide, DTT: dithiothreitol, ESI: electron spray ionisation, FC: flash chromatography, Fmoc: [(9H-fluoren-9-ylmethoxy]carbonyl, Fmoc-OSu: N-({[(9H-fluoren-9-yl)methoxy]carbonyl}oxy)succinimide, GdmCl: guanidinium hydrochloride, HATU: O-(7-azabenzotriazol-1-yl)-1,1,3,3-tetramethyluronium hexafluorophosphate, HOBt: 1-hydroxy-1H-benzotriazole, HPLC: high-performance liquid chromatography, HR: high resolution, h.v.: high vacuum (0.01 – 0.1 Torr), MALDI :matrix-assisted laser desorption ionization, MeIm: 1-methyl-1H-imidazole, MSTN: 1-(mesitylene-2-sulfonyl)-3-nitro-1H-1,2,4-triazole, NMM: N-methylmorpholine,

¹³) See Sect. 7.6 in the review article [25a,b]

¹⁴) The diselenide **11c** is rather poorly soluble in CD₃OH, but the NMR spectrum in this solvent shows sharp signals and well separated N*H* chemical shifts. This would indicate that the β -peptide **11c** is not aggregated. Aggregation normally gives rise to broadened NMR peaks (*cf.* [36])

PMBCl: para-methoxybenzyl chloride, SPS: solid-phase synthesis, TFA: trifluoroacetic acid, (CF₃COOH), TIS: triisopropylsilane ((i-Pr)₃SiH), TNBS: 2,4,6-trinitrobenzene sulfonic acid. Solvents for chromatography and workup procedures were distilled from Sikkon (anh. $CaSO_4$; Fluka) and from KOH (Et₂O). Amino acids were purchased from *Fluka* or *Senn*. CH₂N₂ was prepared according to [40]. All other reagents were used as received from Fluka or Aldrich. FC: Fluka silica gel 60 (40-63 µm), TLC: Merck silica gel $60 F_{254}$ plates; detection with UV, anisaldehyde soln. (9.2 ml anisaldehyde, 12.5 ml conc. H₂SO₄, 3.75 ml AcOH, 340 ml EtOH), or 'Mo-stain' soln. (25 g phosphormolybdic acid, 10 g Ce(SO₄)₂·H₂O, 60 ml conc. H₂SO₄, 940 ml H₂O). Circular Dichroism (CD) Spectra: Jasco J-710 spectropolarimeter from 190-250 nm at 20° in 1-mm rectangular cells, all spectra were corrected for the corresponding solvent spectrum, the molar ellipticity $[\theta]$ in deg \cdot cm² \cdot dmol⁻¹ calculated for the corresponding peptide. IR Spectra: Perkin-Elmer 1600 FT-IR spectrophotometer. NMR Spectra: Bruker AMX-300 (1H: 300 MHz, 13C: 75 MHz), Varian Mercury XL 300 (1H: 300 MHz, 13C: 75 MHz); chemical shifts δ in ppm downfield from internal Me₄Si (=0 ppm); J values in Hz. MS: IonSpec Ultima 4.7 T FT Ion Cyclotron Resonance (ICR; HR-MALDI, in 2,5-dihydroxybenzoic acid matrix) spectrometer; in m/z (% of basis peak). Anal. and prep. RP-HPLC: Merck HPLC system (LaChrom, pump type L-7150, UV detector L-7400, interface D7000, HPLC manager D7000) or Waters HPLC system (pump type 515, PDA-UV detector 996 or dual wavelength detector 2487, HPLC manager Empower Pro). LC/MS was performed on a Thermo Separation Products (TSP) HPLC system connected to a PDA detector (UV6000LB, TSP) and to an ion-trap mass-spectrometry detector (LCQdeca, Finnigan). Linear gradients of 5-60% MeCN in H₂O/0.1% formic acid (0.2 ml/min) on a Waters Atlantis dC18-3 3×100 mm column were used. Elemental analyses were performed by the Microanalytical Laboratory of the Laboratorium für Organische Chemie, ETH-Zürich.

2. Synthesis of the β -Amino-Acids 4 and 7. Fmoc-Sec(PMB)-OH (2) [11b,e]. NaBH₄ (400 mg, 15 mmol) was added in portions to an ice-cooled soln. of selenocystine (1) (500 mg, 1.50 mmol) in 0.5N NaOH soln. (8 ml) and EtOH (2 ml), and the mixture was stirred at r.t. until the yellow color disappeared. After cooling again to 0°, degassed 2N NaOH soln. (4 ml) and PMBCl (0.82 ml, 6.00 mmol) were added. The mixture was stirred 30 min at 0° and 3 h at r.t. Then, the pH was adjusted to 1 – 2 with 6N HCl, and the mixture was kept at 0° overnight. The product was collected by filtration, washed with cold H₂O, and dried under h.v. overnight. Then, a soln. of Fmoc-OSu (972 mg, 2.88 mmol) in MeCN (4 ml) was added to an ice-cooled suspension of the crude product in a mixture of H₂O (7 ml) and Et₃N (0.42 ml, 2.99 mmol). After the addition of Et₃N (0.42 ml, 2.99 mmol), the mixture was stirred at r.t. for 1 h. Then, the pH was adjusted to 1 – 2 with 1N HCl, and the aq. phase was extracted with AcOEt (3 × 30 ml). The combined org. layers were washed with 1N HCl (20 ml) and brine (20 ml), and dried (Na₂SO₄) and concentrated under reduced pressure. FC (SiO₂; hexane/AcOEt/AcOH 25:25:1) yielded **2** (1.173 g, 77%). Anal. data are in accordance with those in [11b].

Boc-Sec(PMB)-OH (5) [11b,e]. NaBH₄ (400 mg, 15 mmol) was added in portions to an ice cooled soln. of 1 (500 mg, 1.50 mmol) in 0.5N NaOH soln. (8 ml) and EtOH (2 ml), and the mixture was stirred at r.t. until the yellow color disappeared. After cooling again to 0°, degassed 2N NaOH soln. (4 ml) and PMBCl (0.82 ml, 6.00 mmol) were added. The mixture was stirred 30 min at 0° and 4 h at r.t. Then, the pH was adjusted to 1-2 and the mixture was kept at 0° overnight. The product was collected by filtration and washed with cold H_2O . After drying under h.v. overnight, the crude product was suspended in MeCN/ sat. aq. NaHCO₃ soln. (20 ml/20 ml), and Boc₂O (753 mg, 3.45 mmol) was added. After stirring overnight citric acid soln. (10% in H_2O , 60 ml) was added, and the aq. phase was extracted with AcOEt (3 × 50 ml). The combined org. layers were washed with brine $(2 \times 50 \text{ ml})$, dried (MgSO₄), and concentrated under reduced pressure. FC (SiO₂; hexane/AcOEt/AcOH 66:33:2) provided 5 (925 mg, 79%). Colorless oil. $R_{\rm f}$ (hexane/AcOEt/AcOH 25:25:1) 0.48. $[a]_{TL}^{tL} = -23.7$ (c = 0.5, MeOH). IR (CHCl₃): 2980m, 1710s, 1610m, 1510s, 1370m, 1035m. 1H-NMR (300 MHz, CD3OD): 1.44 (s, t-Bu); 2.80-2.91 (m, CH2); 3.73 (s, Me); $3.78 (s, CH_2)$; 4.19 (t, J = 6.3, CH); 6.81 (d, J = 8.7, 2 arom. H); 7.20 (d, J = 8.7, 2 arom. H). ¹³C-NMR (75 MHz, CD₃OD): 26.1, 28.0 (CH₂); 29.0 (*Me*₃C); 55.3 (CH); 55.9 (CH₂); 81.0 (C); 115.1, 131.4 (CH); 132.3, 157.6, 159.9, 174.5 (C). ESI-MS: 412.2 (100, [M+D]⁺), 529.9 (23), 304.2 (38). MALDI-HR-MS: 558.0894 ($[M + Na]^+$, $C_{27}H_{25}N_3O_4SeNa^+$; calc. 558.0903).

1-Diazo-3-({[(9H-fluoren-9-yl)methoxy]carbonyl]amino)-4-[(4-methoxybenzyl)selanyl]butan-2one (Fmoc-Sec(PMB)-CHN₂; **3**). Compound **2** (510 mg, 1.00 mmol) was dissolved in THF (5 ml) under Ar, and the mixture was cooled to -25° . After addition of Et₃N (0.15 ml, 1.07 mmol) and ClCO₂ⁱBu (0.14 ml, 1.07 mmol), the mixture was stirred at -25° for 30 min. The resulting white suspension was allowed to warm to 0° , and a soln. of CH₂N₂ in Et₂O was added until the yellow color persisted (ca. 3.00 mmol). Stirring was continued for 4 h as the mixture was allowed to warm to r.t. Excess CH₂N₂ was destroyed by the addition of a few drops of AcOH and vigorous stirring for 5 min. The mixture was diluted with Et₂O (50 ml), and washed with sat. aq. NaHCO₃ soln. (20 ml), sat. aq. NH₄Cl (20 ml) and brine (20 ml). The org. layer was dried (Na₂SO₄) and concentrated under reduced pressure. FC (SiO₂; hexane/AcOEt 2:1 to 1:1) gave 3 (392 mg, 73%). Yellow solid. M.p. $90-92^{\circ}$. $R_{\rm f}$ (hexane/AcOEt 2:1) 0.20. $[\alpha]_{D^{L}}^{\text{tr}} = -36.1 \ (c = 0.8, \text{ CHCl}_3)$. IR (CHCl_3): 3005w, 2110s. 1720s, 1640m, 1610m, 1510s, 1450m, 1365s, 1035m. ¹H-NMR (300 MHz, CDCl₃): 2.70-2.86 (*m*, CH₂); 3.75 (br. s, CH₂, Me); 4.17 (*t*, J = 6.3, CH); 4.32–4.56 (*m*, CH, CH₂); 5.31 (*s*, CH); 5.72 (br. *d*, *J* = 7.4, NH); 6.82 (*d*, *J* = 8.5, 2 arom. H); 7.21 (*d*, J = 8.5, 2 arom. H); 7.30–7.43 (m, 4 arom. H); 7.61 (br. s, 2 arom. H); 7.77 (d, J = 7.4, 2 arom. H). ¹³C-NMR (75 MHz, CDCl₃): 24.8, 27.4 (CH₂); 47.0, 54.3 (CH); 55.0 (Me); 57.1 (CH); 66.6 (CH₂); 113.8, 119.8, 124.9, 126.9, 127.5, 129.8 (CH); 130.3, 141.1, 143.4, 155.5, 158.3, 193.1 (C). ESI-MS: 557.8 (100, [M+Na]⁺), 529.9 (23), 304.2 (38). MALDI-HR-MS: 558.0894 ([M+Na]⁺, C₂₇H₂₅N₃NaO₄Se⁺; calc. 558.0903).

3-([[(9H-Fluoren-9-yl)methoxy]carbonyl]amino)-4-[(4-methoxybenzyl)selanyl]butanoic Acid (Fmoc-β³hSec(PMB)-OH; **4**). Compound **3** (392 mg, 0.73 mmol) was dissolved in THF/H₂O (8 ml/ 2 ml), and PhCO₂Ag (34 mg, 0.15 mmol) was added with exclusion of light. The mixture was ultrasonicated at r.t. After 4 h, H₂O (20 ml) was added, and the pH was adjusted to 1 – 2 with 1N HCl. The aq. phase was extracted with AcOEt (3 × 20 ml), and the combined org. layers were dried (Na₂SO₄) and concentrated under reduced pressure. FC (SiO₂; hexane/AcOEt/AcOH 66:33:2) yielded **4** (273 mg, 71%). White solid. M.p. 120–122°. R_f (hexane/AcOEt/AcOH 66:33:2) 0.42. [α]_D^{TL} = –11.0 (c = 0.5, CHCl₃). IR (CHCl₃): 3010w, 1720s, 1610w, 1510s, 1450w, 1040m. ¹H-NMR (300 MHz, CDCl₃): 2.40–2.83 (m, 2 CH₂); 3.75 (br. s, CH₂, Me); 4.17 (br. s, CH); 4.22 (t, J = 6.5, CH); 4.40 (br. s, CH₂); 5.37 (br. d, J = 9.1, NH); 6.80 (d, J = 7.9, 2 arom. H); 7.20 (d, J = 7.9, 2 arom. H); 7.26 – 7.48 (m, 4 arom. H); 7.59 (d, J = 7.3, 2 arom. H); ¹³C-NMR (75 MHz, CDCl₃): 27.2, 27.7, 37.9 (CH₂); 47.1, 47.8 (CH); 55.2 (Me); 66.8 (CH₂); 113.9, 119.9, 125.0, 127.0, 127.7, 129.9 (CH); 130.6, 141.2, 143.7, 155.7, 158.4, 176.4 (C). MALDI-MS: 548 (100, [M + Na]⁺), 526 (16, [M + H]⁺), 346 (48). MALDI-HR-MS: 548.0938 ([M + Na]⁺, C₂₇H₂₇NNaO₅Se⁺; calc. 548.0947). Anal. calc. for C₂₇H₂₇NO₅Se (524.47): C 61.38, H 5.19, N 2.67; found: C 61.58, H 5.21, N 2.71.

1-Diazo-3-{[(tert-Butoxy)carbonyl]amino}-4-[(4-methoxybenzyl)selanyl]butan-2-one (Boc-Sec(PMB)-CHN₂, 6). Compound 5 (1.55 g, 4.00 mmol) was dissolved in THF (20 ml) under Ar and cooled to -25° . After addition of Et₃N (0.60 ml, 432 mg, 4.27 mmol) and ClCO₂ⁱBu (0.56 ml, 585 mg, 4.28 mmol), the mixture was stirred at -25° for 30 min. The resulting white suspension was allowed to warm to 0° , and a soln. of CH₂N₂ in Et₂O was added until the yellow color persisted (*ca.* 12.0 mmol). Stirring was continued for 4 h as the mixture was allowed to warm to r.t. Excess CH₂N₂ was destroyed by the addition of a few drops of AcOH and vigorous stirring for 5 min. The mixture was diluted with Et₂O (100 ml), and washed with sat. aq. NaHCO₃ soln. (20 ml), sat. aq. NH₄Cl (20 ml), and brine (20 ml). The org. layer was dried (Na₂SO₄) and concentrated under reduced pressure. FC (SiO₂; hexane/AcOEt 2:1) provided **6** (1.36 g, 82%). Yellow oil. $R_{\rm f}$ (hexane/AcOEt 2:1) 0.44. $[\alpha]_{\rm D^{\rm L}}^{\rm re} = -30.4$ (c = 0.9, CHCl₃) IR (CHCl₃): 3010w, 2110s. 1710s, 1640m, 1510s, 1490s, 1370s, 1035m. ¹H-NMR (300 MHz, CDCl₃): 1.43 (s, t-Bu); 2.76 (*d*, *J* = 6.0, CH₂); 3.76 (*m*, CH₂, Me); 4.34 (br. *s*, CH); 5.33, 5.48 (2*s*, CH, NH); 6.81 (*d*, *J* = 8.4, 2 arom. H); 7.20 (d, J = 8.4, 2 arom. H). ¹³C-NMR (75 MHz, CDCl₃): 25.1, 27.4 (CH₂); 28.2 (Me₃C); 54.2 (CH); 55.1 (Me); 56.8 (CH); 80.1 (C); 113.9, 129.9 (CH); 130.5, 155.0, 158.4, 192.7 (C). ESI-MS: 436.1 $(100, [M + Na]^+), 414.2 (6). MALDI-HR-MS: 436.0742 ([M + Na]^+, C_{17}H_{23}N_3NaO_4Se^+; calc. 436.0746).$

3-{[(tert-Butoxy)carbonyl]amino]-4-[(4-methoxybenzyl)selanyl]butanoic Acid (Boc- β^3 hSec(PMB)-OH, **7**). Compound **6** (1.24 g, 3.00 mmol) was dissolved in THF/H₂O (32 ml/8 ml), and PhCO₂Ag (137 mg, 0.60 mmol) was added with exclusion of light. The mixture was ultrasonicated at r.t. After 4 h, H₂O (20 ml) and citric acid soln. (10% in H₂O, 30 ml) were added, and the aq. phase was extracted with AcOEt (1 × 100 ml, 2 × 50 ml). The combined org. layers were washed with brine (40 ml), dried (Na₂SO₄), and concentrated under reduced pressure. FC (SiO₂; hexane/AcOEt/AcOH 25:25:1) gave **7** (1.06 g, 88%). Yellow oil, which slowly crystallized to a pale yellow solid. M.p. 85–87°. R_f (hexane/

AcOEt/AcOH 25:25:1) 0.54. $[\alpha]_{L^1}^{p_1} = +8.5$ (c = 0.65, CHCl₃). IR (CHCl₃): 3010m, 1710s, 1610w, 1510s, 1370m, 1035m. ¹H-NMR (300 MHz, CDCl₃): 1.43 (s, t-Bu); 2.50-2.79 (m, 2 CH₂); 3.75 (br. s, CH₂, Me); 4.07 (br. s, CH); 5.14 (br. s, NH); 6.81 (d, J=8.6, 2 arom. H); 7.19 (d, J=8.6, 2 arom. H). ¹³C-NMR (75 MHz, CD₃OD): 21.8 (CH₂); 28.4 (Me₃C); 29.4, 40.1 (CH₂); 49.7 (CH); 55.9 (Me); 80.4 (Me₃C); 115.1, 131.4 (CH); 132.9, 157.8, 160.1, 175.0 (C). MALDI-MS: 426.1 (100, [M+Na]⁺), 404.1 (16, [M+ H]⁺). MALDI-HR-MS: 426.0783 ($[M + Na]^+$, $C_{17}H_{25}NNaO_5Se^+$; calc. 426.0790). Anal. calc. for $C_{17}H_{25}NO_5Se$ (402.35): C 50.75, H 6.26, N 3.48; found: C 50.96, H 6.34, N 3.51.

Pentafluorophenyl 3-{[(tert-Butoxy)carbonyl]amino]-4-[(4-methoxybenzyl)selanyl]butanoate (Boc- $\beta^3hSec(PMB)$ -OPFP; **7a**) [35]. To a stirred ice-cooled soln. of **7** (402 mg, 1.00 mmol) and pentafluorophenol (202 mg, 1.10 mmol) in dry THF (10 ml) DCC (227 mg, 1.1 mmol) was added. Stirring was continued for 1 h at 0° and for 1 h at r.t. The suspension was cooled in an ice bath, filtered, and the filtrate was washed with cold THF. The solvent was evaporated. The crude product was treated with Et₂O (60 ml), filtered, and the filtrate was evaporated to dryness. Recrystallization of the crude product from hexane yielded **7a** (411 mg, 72%). Light yellow solid. M.p. 108° . R_f (hexane/AcOEt 4:1) 0.40. $[a]_{TL}^{TL} = 3.6$ $(c = 0.5, CHCl_3)$. IR (CHCl_3): 2980w, 1785m, 1710m, 1610w, 1520s, 1515m, 1370m, 1100m, 1005m. ¹H-NMR (300 MHz, CDCl₃): 1.46 (s, t-Bu); 2.69–2.80 (m, CH₂); 2.95–3.06 (m, CH₂); 3.78 (s, Me); 3.81 (*s*, Me); 4.21 (br. *s*, CH); 4.90 (br. *s*, NH); 6.82 (*d*, *J* = 8.7, 2 arom. H); 7.23 (*d*, *J* = 8.7, 2 arom. H). ESI-MS: 591.9 (100, $[M + Na]^+$). Anal. calc. for $C_{23}H_{24}F_5NO_5Se$ (568.40): C 48.60, H 4.26, N 2.46; found: C 48.33, H 4.26, N 2.53.

3. Peptide Synthesis. 3.1. Fmoc Deprotection. General Procedure 1 (GP 1). The Fmoc group was removed by treating the resin with 20% piperidine in DMF (4 ml, 3×10 min) under Ar bubbling. After filtration, the resin was washed with DMF (5 ml, 4×1 min).

3.2. Coupling of the β -Amino Acids on Rink-Amide AM Resin. General Procedure 2 (GP 2). GP 2a: The Fmoc deprotection was carried out according to GP1. Solid-phase synthesis was continued by sequential incorporation of Fmoc-protected amino acids. For each coupling step, the resin was treated with a soln. of the Fmoc-protected β^3 -amino acid (3 equiv.), HATU (2.8 equiv.), and DIPEA (6 equiv.) dissolved in the minimal amount of DMF for 1 h. Monitoring of the coupling reaction was performed with the TNBS test [41]. In the case of a positive TNBS test (indicating incomplete coupling), the suspension was allowed to react further for 1 h, or, after filtration, the peptide-resin was treated again with the same Fmoc-protected amino acid (1-3 equiv.), and with the coupling reagents. The resin was then filtered and washed with DMF (20 ml/g resin, 5×1 min) prior to the following deprotection step. GP 2b: As described in GP 2a, but without Fmoc deprotection (GP 1).

3.3. Rink-Amide AM Resin Cleavage and Final Deprotection. General Procedure 3 (GP 3). The cleavage from the resin and the peptide deprotection were performed according to [42]. The dry peptideresin was suspended in a soln. of TFA/H₂O/CH₂Cl₂/TIS 89:5:5:1 (10 ml) for 2 h. The resin was removed by filtration, washed with TFA ($2 \times$), and the org. phase was concentrated under reduced pressure. The resulting oily residue was treated with cold Et₂O, and the formed precipitate was separated. The isolated crude β -peptide was dried under h.v. and stored at -20° before purification.

3.4. HPLC Analysis and Purification of β -Peptides. General Procedure 4 (GP 4). RP-HPLC Analysis was performed on a Macherey-Nagel C_{18} column (Nucleosil 100-5 C_{18} (250 × 4 mm)) by using a linear gradient of A: 0.1% TFA in H₂O and B: MeCN, at a flow rate of 1 ml/min with UV detection at 220 nm. $t_{\rm R}$ in min. Crude products were purified by prep. RP-HPLC on a Macherey-Nagel C_{18} column (Nucleosil $100-7 C_{18} (250 \times 21 \text{ mm})$ using gradient of A and B at a flow rate of 17 ml/min with UV detection at 220 nm, and then lyophilized.

 $H-\beta^3hSec(PMB)-\beta^3hAsp-\beta^3hSer-\beta^3hPhe-\beta^3hLys-\beta^3hThr-\beta^3hVal-NH_2$ (8). Rink-Amide AM resin (282 mg, 0.2 mmol, loading 0.71 mmol/g) was preswelled in DMF (10 ml) for 1 h by N₂ bubbling. Subsequently the resin was filtered. SPS was performed according to GP 2a by sequential incorporation of Fmoc- $\beta^{3}hVal-OH$, Fmoc- $\beta^{3}hThr(\beta u)-OH$, Fmoc- $\beta^{3}hLys(Boc)-OH$, Fmoc- $\beta^{3}hPhe-OH$, Fmoc- $\beta^{3}hSer-\beta^$ OH, and Fmoc-*β*³hAsp('Bu)-OH. Then, the resin was Fmoc-deprotected (*GP1*), washed with CH₂Cl₂ (4 ml, 4 \times 1 min), and dried under h.v. overnight. The dried resin was divided in two equal parts (2 \times 221 mg, 2×0.10 mmol).

The first part (221 mg, 0.10 mmol) was preswelled in DMF (10 ml) for 1 h by N₂ bubbling. The resin was filtered and Fmoc- β^3 hSec(PMB)-OH was coupled according to GP 2b. Then the resin was washed with CH₂Cl₂ (4 ml, 4 × 1 min) and dried under h.v. overnight. Treatment of the peptide-resin according to *GP 3* afforded the crude peptide **8** (168 mg) as a TFA salt. Purification of this crude peptide by RP-HPLC (30–95% *B* in 30 min: 5 min 30% *B*; 25 min 40% *B*; 30 min 95% *B*; $t_{\rm R}$ 17.1 min) according to *GP 4* yielded the TFA salt of **8** (27.1 mg) in >98% purity.

The second part (221 mg, 0.10 mmol) was preswelled in DMF (10 ml) for 1 h by N₂ bubbling. After filtration, a mixture of Fmoc- β^3 hSec(PMB)-OPFP (**7a**) (171 mg, 0.30 mmol) and HOBt (41 mg, 0.30 mmol) in DMF was added to the resin. The suspension was mixed by N₂ bubbling for 2 h. Subsequently, the resin was filtered, washed with DMF (4 ml, 4 × 1 min), with CH₂Cl₂ (4 ml, 4 × 1 min), and dried under h.v. overnight. Treatment of the peptide-resin according to *GP* 3 afforded the crude peptide **8** (127 mg) as a TFA salt. Purification of this crude product by RP-HPLC (30–95% *B* in 30 min: 5 min 30% *B*; 25 min 40% *B*; 30 min 95% *B*; t_R 17.1 min) according to *GP* 4 yielded the TFA salt of **8** (24.7 mg) in >98% purity.

 $\label{eq:RP-HPLC} \begin{array}{l} (5-99\% \ B \ in \ 60 \ min: 45 \ min \ 50\% \ B, \ 60 \ min \ 99\% \ B): t_{R} \ 33.79, \ purity \ > 98\%. \ MALDI-HR-MS: \ 1064.5281 \ ([M+H]^+, \ C_{49}H_{78}N_9O_{12}Se^+; \ calc. \ 1064.5299). \end{array}$

 $[H-\beta^3hSec-\beta^3hAsp-\beta^3hSer-\beta^3hPhe-\beta^3hLys-\beta^3hThr-\beta^3hVal-NH_2]_2$ (9). Compound 8 (25.8 mg, 20 µmol) was dissolved in 1 ml of a mixture of *m*-cresol, thioanisol, TFA, and TMSBr 50:120:750:132 under Ar in an ice bath. After stirring for 1 h, cold Et₂O was added, and the precipitate formed was washed with cold Et₂O (3×). This crude product was dried under h.v. Purification of the crude peptide by RP-HPLC (20–95% B in 70 min: 5 min 20% B; 65 min 50% B; 70 min 95% B; 17 ml/min) yielded the TFA salt of 9 (13.9 mg, 59%; t_R 25.8 min) and the TFA salt of 8 (4.7 mg, 18%, t_R 37.9 min).

RP-HPLC (5−99% *B* in 60 min: 45 min 50% *B*, 60 min 99% *B*): t_R 28.11, purity >98%. MALDI-HR-MS: 1885.855 ($[M+H]^+$, $C_{82}H_{136}N_{18}O_{22}Se_2^+$; calc. 1885.8508).

3. Selenoligation. $[H-\beta^3hAla-\beta^3hPhe-\beta^3hTyr-\beta^3hGly-\beta^3hSec-\beta^3hAsp-\beta^3hSer-\beta^3hPhe-\beta^3hLys-\beta^3hThr \beta^3hVal-NH_2J_2$ (11c) and H- β^3hAla - β^3hPhe - β^3hTyr - β^3hGly - $\beta^3hSec(SPh)$ - β^3hAsp - β^3hSer - β^3hPhe - β^3hLys - $\beta^3 hThr-\beta^3 hVal-NH_2$ (11b). Degassed buffer (2.0 ml, 100 mM phosphate, 6M guanidinium HCl) was added to a mixture of the peptide 10 (3.0 mg, 4.06 µmol) and the diselenide of 9 (5 mg, 2.14 µmol) in a Schlenk flask under N_2 . The reaction was initated by the addition of 5% thiophenol (100 μ l). The heterogeneous mixture was vigorously stirred at r.t. Aliquots were withdrawn, diluted 10 times, and then analyzed by HPLC and LC/MS. After the reaction was judged to be over, the mixed selenosulfide was reduced with DTT (10 mg, 65 µmol) for 30 min, and then it was transferred to a 15-ml Falcon tube and extracted with Et₂O (2 × 10 ml). The resulting soln. was diluted with 3 ml of 50% MeCN in H₂O (0.1% TFA), and that was directly purified by prep. HPLC. The desired product was isolated as the diselenide of product 11c in 50% yield (3.4 mg). RP-HPLC (5–95% B in 60 min: 5 min 5% B, 50 min 60% B, 55 min 95% B): $t_{\rm R}$ 35.94, purity > 99%. MALDI-HR-MS for the [A + 12] peak: 2874.365 ($[M + H]^+$; calc. for $C_{136}H_{205}N_{26}O_{32}Se_2^+: 2874.359$). Following an alternative workup where Et₂O was omitted and the crude mixture was directly injected onto the RP-HPLC, the product peak contained a large amount of thiophenol in addition to 11b. Most of the thiophenol contaminant could be removed by lyophilization, allowing isolation of **11b** in 66% yield (5 mg). RP-HPLC as for **11c**: $t_{\rm R}$ 36.74, purity 90%. LC/ESI-MS: 1546.8 ($[M+H]^+$, $C_{74}H_{108}N_{13}O_{16}SSe^+$; calc. 1546.7).

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